

4-(2-Nitrobenzylideneamino)-3-(1*H*-1,2,4-triazol-1-ylmethyl)-1*H*-1,2,4-triazole-5(4*H*)-thioneXiao-Lan Lu,<sup>a</sup> Wen-Zhao Bi,<sup>b</sup>  
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Xu<sup>b\*</sup> and Guan-Ping Yu<sup>b</sup><sup>a</sup>College of Chemistry and Chemical Engineering, Ocean University of China, Qingdao 266003, People's Republic of China, and <sup>b</sup>College of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, Qingdao 266042, People's Republic of China

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## Key indicators

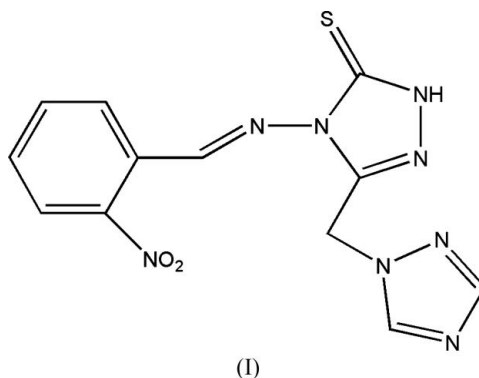
Single-crystal X-ray study  
*T* = 294 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
*R* factor = 0.043  
*wR* factor = 0.112  
Data-to-parameter ratio = 14.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound,  $\text{C}_{12}\text{H}_{10}\text{N}_8\text{O}_2\text{S}$ , the thione-substituted triazole ring forms dihedral angles of 79.86 (2) and 9.86 (3)°, respectively, with the other triazole ring and the benzene ring. Intermolecular  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds link the molecules into chains extended along the [101] direction. The crystal packing is further stabilized by van der Waals forces.

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## Comment

Recently, compounds containing the 1*H*-1,2,4-triazole group have attracted much interest because they have good plant-growth regulatory activity for a wide variety of crops (Xu *et al.*, 2002). In addition, amine- and thione-substituted triazoles have been studied as anti-inflammatory and antimicrobial agents (Eweiss *et al.*, 1986; Awad *et al.*, 1991). In a search for new triazole compounds with better biological activity, the title compound, (I), was synthesized. We report here the crystal structure of (I) (Fig. 1).



Bond lengths and angles in the triazole rings in (I) are in agreement with those in previous reports (Li *et al.*, 2005; Xu *et al.*, 2005). The molecule exists in the thione tautomeric form, with an  $\text{S}=\text{C}$  distance of 1.672 (2) Å, indicating substantial double-bond character (Escobar-Valderrama *et al.*, 1989). The mean planes  $\text{C}10-\text{C}12/\text{N}6/\text{N}7/\text{N}8$  and  $\text{C}1-\text{C}7/\text{N}1$  make dihedral angles of 79.86 (2) and 9.86 (3)°, respectively, with the thione-substituted triazole plane  $\text{C}8-\text{C}10/\text{N}3/\text{N}4/\text{N}5/\text{S}1$ .

Intermolecular  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds (Table 1) link the molecules into chains extended along the [101] direction. The crystal packing (Fig. 2) is further stabilized by van der Waals forces.

## Experimental

A mixture of 4-amino-3-(1,2,4-triazole-1-yl)-1*H*-1,2,4-triazole-5(4*H*)-thione (5 mmol) and 2-nitrobenzaldehyde (5 mmol) was refluxed for

10–20 min in glacial acetic acid. The mixture was then filtered and crystallized from ethanol to afford the title compound (yield: 1.55 g, 93.7%). Single crystals suitable for X-ray measurements were obtained by recrystallization from ethanol at room temperature.

Crystal data

$C_{12}H_{10}N_8O_2S$   
 $M_r = 330.34$   
 Monoclinic,  $P2_1/n$   
 $a = 11.421 (2) \text{ \AA}$   
 $b = 8.1070 (14) \text{ \AA}$   
 $c = 16.446 (3) \text{ \AA}$   
 $\beta = 105.345 (3)^\circ$   
 $V = 1468.5 (5) \text{ \AA}^3$

$Z = 4$   
 $D_x = 1.494 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.25 \text{ mm}^{-1}$   
 $T = 294 (2) \text{ K}$   
 Block, yellow  
 $0.18 \times 0.10 \times 0.06 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.950, T_{\max} = 0.985$

8026 measured reflections  
 2991 independent reflections  
 1894 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.038$   
 $\theta_{\text{max}} = 26.4^\circ$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.112$   
 $S = 1.03$   
 2991 reflections  
 212 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0393P)^2 + 0.548P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.26 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N4-H4A\cdots N8^i$	0.88 (3)	1.97 (3)	2.838 (3)	171 (2)

Symmetry code: (i)  $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$ .

The positional and isotropic displacement parameters of the H atoms attached to N4 were refined freely. All other H atoms were placed in calculated positions, with  $C-H = 0.93$  or  $0.97 \text{ \AA}$ , and refined using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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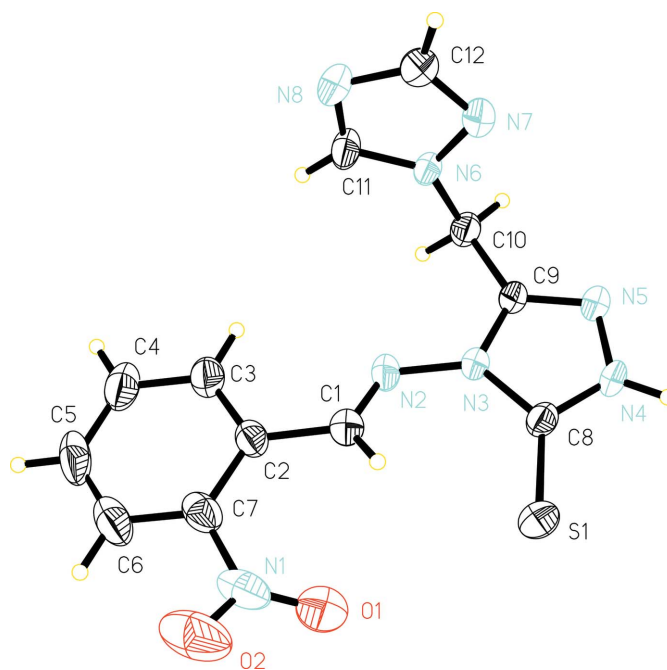


Figure 1 View of (I), with displacement ellipsoids drawn at the 40% probability level.

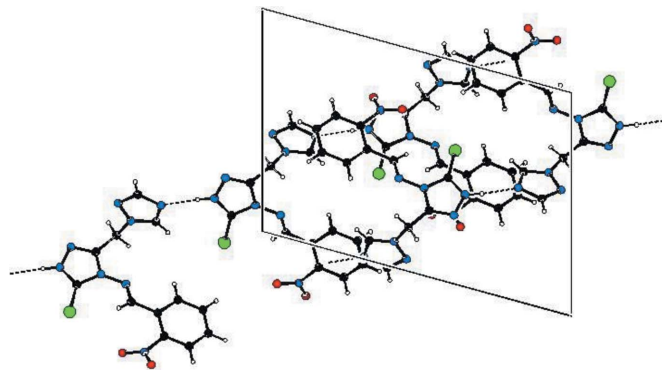


Figure 2 A packing diagram, viewed down the  $b$  axis. Hydrogen bonds are shown as dashed lines.

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